Aldehyde and phosphinate analogs of glutathione and glutathionylspermidine: potent, selective binding inhibitors of the E. coli bifunctional glutathionylspermidine synthetase/amidase

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Introduction: The tripeptide glutathione is converted to glutathionylspermidine (Gsp) in *Escherichia coli* and in trypanosomatid parasites by an ATP-cleaving Gsp synthetase activity. In parasites, an additional glutathionylation forms bis- (glutathionyl)-spermidine, trypanothione, believed to be the major surveillance thiol involved in oxidant defense mechanisms in trypanosomatid parasites. In E. coli, the Gsp synthetase is part of a bifunctional enzyme opposed by the hydrolytic Gsp amidase.

Results: Gsp amidase and Gsp synthetase activities of the bifunctional E . coli enzyme can be separately targeted by potent, selective slow-binding inhibitors that induce time-dependent inhibition. The inhibitor y-Glu-Ala-Gly-CHO most probably captures Cys59 and accumulates as the tetrahedral adduct in the amidase active site. Inhibitory Gsp phosphinate analogs are phosphorylated by ATP to yield phosphinophosphate analogs in the synthetase active site. Binding of phosphinophosphate in the Gsp synthetase active site potentiates the inhibition affinity for the aldehyde at the Gsp amidase active site by two orders of magnitude.

Conclusions: Time-dependent inhibition of the Gsp amidase activity by the aldehyde substrate analog supports previous work that suggests glutathionyl acyl-enzyme intermediate formation in the Gsp amidase reaction mechanism. Use of potent selective inhibitors against Gsp amidase (aldehyde) and Gsp synthetase (phosphinate) activities provides further evidence of interdomain communication in the bifunctional enzyme from E. coli.

Introduction

In *Escherichia coli* and in trypanosomatid parasites, the tripeptide thiol glutathione (GSH) is enzymatically converted to N^1 -glutathionylspermidine (Gsp) by an ATPcleaving, amide-forming Gsp synthetase [l-4]. In the parasites, a subsequent glutathionylation at the N^8 of Gsp leads to $N^{1,8}$ -bis(glutathionyl)-spermidine, known as trypanothione [1,5]. When we purified and characterized the 70 kDa E. coli Gsp synthetase, a second activity, hydrolysis of Gsp back to GSH and spermidine (Figure l), was detected $[3]$. We confirmed that the E. coli enzyme is a bifunctional protein with two separate catalytic domains by expressing an amino-terminal 25 kDa fragment (residues l-225) as an active Gsp amidase and residues 189 to 619 as a carboxyl-terminal Gsp synthetase fragment [6]. Studies to probe both the regulation of the two opposing activities (to avoid a futile GSH-dependent ATPase cycle) and mechanistic analyses of both the synthetase and amidase catalytic mechanisms have been undertaken.

Interest in the Gsp synthetase/amidase from E. co/i mercat in the Osp synthetase/annually nome in

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in trypanothione biosynthesis, a Gsp synthetase and trypanothione synthetase. As in the E . coli Gsp synthetase/amidase, both trypanosomal enzymes catalyze amide bond formation coupled to ATP hydrolysis. Because trypanothione represents the primary oxidant defense mechanism in trypanosomes, the unique trypanothione synthetase represents an attractive drug target for the treatment of trypanosomiasis, which includes the significant medical problems of South American Chagas' disease and African sleeping sickness.

 $E.$ coli, however, utilizes a GSH-based system for oxidant defense; thus the physiological role of the enzyme is unclear. Synthesizing Gsp from GSH and spermidine, the Gsp synthetase/amidase represents an intriguing juxtaposition of two important classes of molecule. Whether participating in oxidant defense mechanisms or in complexation with nucleic acids, phospholipids, or proteins via spermidine, the intrinsic balance and regulation of synthetase and amidase activi- $\frac{1}{2}$ $\frac{1}{2}$ function $\frac{1}{2}$

In a preliminary communication, we reported that the Gsp phosphinate analog, la (Figure Z), is a potent time-dependent inhibitor that selectively inactivates the synthetase in the presence of ATP but does not inhibit the amidase activity of the bifunctional enzyme [7]. Failure of la to inhibit the Gsp amidase is inconsistent with a mechanism involving direct attack of $H₂O$, since the phosphinate should be a good mimic of such a tetrahedral adduct. Rather, lack of Gsp amidase inhibition by the phosphinate la suggests initial attack by an enzyme nucleophile. Indeed, presteady-state kinetic analysis with the chromogenic y-Glu-Ala-Gly-0-p-nitrophenyl ester substrate gave burst data consistent with rate-determining breakdown of an acyl enzyme intermediate [8]. The amidase is not inhibited by a generic serine protease inhibitor (3,4-dichloroisocoumarin) but is inactivated by iodoacetamide. Furthermore, mutagenesis of Cys59 to Ala inactivates the Gsp amidase [8]. As a further test of the candidacy of Cys59 as a catalytic nucleophile in Gsp amidase activity, we report the synthesis of the y-Glu-Ala-Gly aldehyde, 2, and its ability to serve as a selective, slow-binding inhibitor of the Gsp amidase activity, consistent with the ability of aldehyde analogs of cysteine protease substrates to form tight long-lived inhibitor complexes [9,10]. We also report the preparation and characterization of serine (lb) and cysteine (lc) analogs of the Ala-containing phosphinopeptide, la, (Figure 2) as slowbinding inhibitors of the Gsp synthetase activity.

Results

Slow binding inhibition of E . coli Gsp amidase by y-Glu-Ala-GlyCHO

Synthesis of the aldehyde analog, 2, of the substrate G is G in G in G in the phoso- G p -ord- n a-ory is summarized in Figure of As in the phos- μ ² complications of this case of the minimum chemistry, in particular, in particular, intermolecular, intermolecular, in complications of thiol chemistry, in particular, intermolecular thiohemiacetal formation between two molecules of 2. As for the reduction of the protected methyl ester 3, excess (four equivalents) diisobutyl aluminum hydride (DIBAL) was used, as the metal hydride would be consumed by amide hydrogens. Furthermore, immediately after the reduction had been accomplished, the final deprotection by trifluoroacetic acid was conducted under argon to prevent oxidation of the aldehyde to the carboxylic acid. After purification by gel filtration and high performance liquid chromatography (HPLC), the target compound 2 was stored at -20° C, and no significant change was observed over a period of a month.

The GSH analog γ -Glu-Ala--Gly is an excellent substrate for the Gsp synthetase activity [6], whereas the corresponding aldehyde, 2, is incompetent as anticipated. In contrast, when aldehyde, 2, was incubated with the pure 25 kDa Gsp amidase fragment, time-dependent inhibition was observed (Figure 4a). With aldehyde 2 at 100 μ M, all activity was lost in a progressive fashion. This pattern is typical for aldehyde analogs of substrates reacting at the active site of cysteine proteases, in which an initial collisional complex (E-1) isomerizes to a tighter complex $(E \cdot I^*)$ according to equation 1. Fitting the observed data to this model allowed determination of K_i and K_i^* (Figure 4b), 1.7 μ M and 80 nM, respectively (Table 1). In order to assess the reversibility of inhibition on dilution, $4 \mu M$ Gsp amidase fragment was incubated with 1 mM aldehyde 2 for 20 min at 37°C to fully

The synthesis of y-Glu-Ala-Gly aldehyde. EDAC, 1 -ethyl-3-3-dimethylaminopropyl) carbodiamide; TFA, trifluoroacetic acid.

Figure 4

(a) inne-dependent immonion of Cisp annuase inagine. By γ -Glu-Ala-Gly aldehyde. (b) Secondary plot of (a). V_s = steady-state rate and V_o = initial rate of curves in (a).

inactivate the enzyme, then diluted 100-fold into a reaction monitor the recovery of activity. A $t_{1/2}$ of 2 min, correspondmixture containing 36 mM γ -Glu-Ala-Gly-p-nitro-aniline to ing to a k_{remain} (rate of regain of activity from fully inhibited state) of 0.35 min⁻¹, was observed (data not shown).

$$
E + I \quad \xrightarrow{\text{K}_{i}} \quad E \cdot I \quad \xrightarrow{\text{K}_{a}} \quad E \cdot I^* \tag{1}
$$

Slow-binding inhibition of E. coli Gsp synthetase by phosphinate analogs

Synthesis of la along with its precursor 5a was carried out as previously described (Figure 5) (S.C. and J.K.C., unpublished data). Compounds Sb and 5c were prepared using similar procedures. However, attempts to convert 5b to 1b, as described for the conversion of 5a to la (treatment of 5a with 30% HBr in AcOH), resulted in the simultaneous acetylation of the serine-hydroxyl group. Consequently, conversion of 5b to lb was achieved by using trimethylsilyl bromide (TMSBr) as the deprotective reagent. Thus, treatment of 5b with excess TMSBr followed by hydrogenolysis gave lb in 52% yield after ion-exchange chromatography purification. The conversion of 5c to 1c proved to be more problematic. Treatment of 5c with either HBr/AcOH or TMSBr resulted in a mixture of products as monitored by 31P nuclear magnetic resonance (NMR), and none of the desired product lc could be isolated. Therefore, 5c was $\frac{1}{1}$, $\frac{1}{2}$ the corresponding cystine-derivative 6 in satisfactory yield. The cysteine-phosphinate 1c could then be generated in situ by the treatment of 6 with dithiothreitol (DTT) prior to enzyme assay. In order to evaluate the polyamine binding requirement during the inhibitor-enzyme interaction, a non-polyamine containing phosphinopeptide was also syn-
thesized in five steps from N-Z-phosphinoglycine [12].

Table 1

In a previous preliminary report we noted that a series of Inhibition of the full-length Gsp synthetase activity by $1a$ Gep phosphapeptide analogs required both a spermidine yielded a K of 3.2 PM and a K $*$ of 7.8 nM (Table 1), a 410moiety and the P-CH-phosphinate linkage in order to fold gain in potency by K, K,* ratio, in accord with equamoiety and the P-CH₂-phosphinate linkage in order to fold gain in potency by K_i/K_i^* ratio, in accord with equa-
observe time-dependent slow-binding inhibition of the tion 1. No inhibition of the amidase was detected [Gsp synthetase activity in full-length bifunctional enzyme. Here, we report the evaluation of Ser (lb) and Cys (lc)

Figure 5

 0.4 0_{nM} 0.3 Absorption (AU) 150 nM 0.2 250 nM $0.1 + 350 \text{ nM}$ - 500 nM 1 u.M 20 u.M $0 \rightarrow 0$ 0 5 10 15 20 25 30 Time (min) Chemistry&Biology

analogs of the Gsp-phosphinate inhibitor, the latter most closely approximating the y-Glu-Cys dipeptidyl moiety of the physiological substrate. Both lb (data not shown) and lc (Figure 6) exhibit time-dependent progressive tightening of inhibition, as observed with la. Phosphinates lb and lc show a 500-fold and 670-fold tightening (Table l), respectively, from low micromolar K_i to low nanomolar K_i^* values upon phosphorylation in the synthetase active site. The Cys-phosphinate 1c ($K_i[*] = 3.1$ nM) forms a threefold tighter complex than either the Ala or Ser analogs, la and **1b**. The $t_{1/2}$ for regain of catalytic activity by dissociation of 1a, b, or c from the enzyme following dilution of the $E-I^*$ complex into assay solutions free of inhibitor was 25, 27 and 28 min, respectively.

Altered potency and equilibrium of y-Glu-Ala-Gly-CHO in full-length Gsp synthetase/amidase - modulation by Gsp phosphinate

The susceptibility of the amidase active site to inhibition by 2 was then examined in the full-length bifunctional enzyme. Time-dependent progressive inhibition was again observed (Figure 7a) with elevated inhibition constants, $K_i = 98 \mu M$ and $K_i^* = 20 \mu M$ (Table 1), reflecting a diminished potency of 50-fold in K_i and 240-fold in K_i^* for the $\frac{1}{2}$ and $\frac{1}{2}$ full-length bifunc-length bifunc-length bifunc-length bifuncannuase active site when comained in functional bifunctional enzyme. We have demonstrated that the k_{est} for amidase activity is suppressed 20-fold in the full-length Gsp synthetase/amidase compared to the Gsp amidase fragment alone $[6]$ and that the phosphinate analog **1a**, binding at the arone to and that the phosphimate analog \mathbf{a} , officing at the symmetries site, accelerates annuase turnover $(\mathbf{k}_{\text{cat}})$ by 20 fold [8]. Consequently, we have examined inhibition of Gsp amidase activity in the full-length enzyme by 2 in the presence of 50 μ M 1a and 2 mM ATP, preincubated for 5 min to saturate the synthetase active site. Comparison of Figure 7b with Figure 7a, shows more potent inhibition by

Figure 6 **Figure 7** Figure 7

(a) The time-dependent inhibition of full-length Gsp synthetase/amidase by y-Glu-Ala-Gly aldehyde. High signal-to-noise ratios due to high background absorbance of the assay substrate, y-Glu-Ala-Gly-p-nitro-aniline, at 36 mM. (b) The time-dependent inhibition of full-length Gsp synthetase/amidase by y-Glu-Ala-Gly aldehyde in the presence of Gsp phosphinate 1a and ATP.

2 when the synthetase active site was saturated by **la** + ATP. A K_i of 6.8 μ M and K_i^{*} of 0.25 μ M were calculated (Table 1). Thus, K_i tightens by 16-fold and K_i* by 80fold in the amidase active site when the synthetase active site is occupied by its slow-binding inhibitor.

Discussion

The tripeptide aldehyde 2 was prepared to test as a slowbinding, time-dependent inactivator of the Gsp amidase activity. Validation of such behavior further supports a activity. randamoli of such behavior further supports a the action incontains in involving indetermine catalysis by $\frac{1}{2}$ annuase, while $\frac{1}{2}$ is $\frac{1}{2}$ as reading candidate for the nactiophile, given cannel results from CDM indiagenesis, iodoacetamide alkylative inactivation, and burst kinetics supporting acyl-enzyme intermediate formation. Other peptide aldehydes have been potent active-site titrants of proteases using cysteine nucleophiles in catalysis, notably
Ac DVED CHO and Ac YVAD CHO for the caspases

interleukin-converting enzyme (ICE) and CPP-32, which are involved in programmed cell death pathways [13-15]. In those cases, the time-dependent inhibition has been correlated with covalent tethering of the aldehyde as the tetrahedral thiohemiacetal adduct to the active cysteine by X-ray structures for both ICE and CPP-32 [15]. An analogous accumulating tetrahedral adduct would be consistent for the Gsp amidase mechanism (equation 2). While Cys 59 is the likely nucleophile, until structural studies prove the point, there is the formal possibility that $Enz-X$ is an oxygen nucleophile, given the precedent that amino-terminal threonine in the proteasome β subunits is trapped as a stabilized hemiacetal [16-181.

$$
R \times R
$$

There is a clear difference in both absolute affinity and partition between $E-I$ and $E-I^*$ of the Gsp amidase active site when in the bifunctional full-length enzyme or when present as the autonomous 25 kDa domain. The SO-fold weaker K_i for the aldehyde 2 in full-length Gsp synthetase/amidase compared to the Gsp amidase fragment suggests an altered ground state geometry. Additionally, the isomerization of initial E.1 to tightened E.1" complex has a 1:4 partition $E \cdot I/E \cdot I^*$ for full-length enzyme, but a 1:20 partition favoring E.1" accumulation in the Gsp amidase fragment. If indeed E.I* is the tetrahedral (thiohemiacetal) adduct of equation 2, the differential stability could reflect subtle differences in microenvironment, such as the pK_a of the attacking enzyme nucleophile or the strength of a hydrogen bond that may stablize the tetrahedral oxyanion.

The second class of Gsp synthetase/amidase inhibitors that show progressive time-dependent inhibition is the series of phosphinate analogs (1a-c), that are selective for the Gsp synthetase active site. Tight-binding properties of these Gsp phosphinate analogs, K/K ^{*} ratios of 410–670, permitted mechanism-based titration of the neighboring permitted meentiment baced attribute of the biographical g the synthetase active site in the bituittional Osp syn thetase/amidase and allowed a test of intersite communication. Preincubation was conducted with $1a + ATP$ to inactivate the Gsp synthetase active site and to load it $max₁$ and $max₂$ dissociation phosphate and $max₁$ dissociation phosphate and $max₁$ with the slowly dissociating phosphinophosphate analog (equation 3) of the tetrahedral reaction intermediate for Gsp synthetase catalysis. Consequently, the Gsp amidase active site could then be probed for K_i and K_i^* for alde-
hyde 2 when the synthetase site was thus occupied.

Indeed, a 16-fold tightening in K_i and, more impressively, an 80-fold tightening in K_i ^{*} were obtained. This result is consistent with the hypothesis of interacting active sites, as we have reported a 20-fold increase in Gsp amidase k_{cat} in the full-length enzyme via the same occupancy of la + ATP at the Gsp synthetase active site or by liberation of the amidase as a separate catalytic fragment [6]. The net K_i^* of 250 nM for the aldehyde 2 when the synthetase active site is occupied by 1a is only threefold higher than the K_i^* of 80 nM of aldehyde 2 for the separate Gsp amidase fragment. Thus, both by analysis of catalytic flux (amidase k_{cat}) or by inhibition potency and K_i/K_i^* equilibria for the aldehyde 2, there is parallel evidence for negative regulation of the upstream Gsp amidase by the downstream Gsp synthetase domain when the bifunctional enzyme is at rest. Furthermore, when the Gsp synthetase domain is in mid-catalytic cycle, its architecture must reversibly alter not only to convert glutathione, ATP and spermidine substrates to produce Gsp, ADP and P_i , but also to transiently release the autoinhibition of the hydrolytic Gsp amidase domain. It remains to be seen whether the k_{cat} effects and K/K ^{*} changes monitor a common set of microscopic steps in amidase catalytic efficiency, for example, by lowering the pK_a of Cys 59 to increase its nucleophilicity in order to facilitate acyl-Senzyme formation (catalysis) or by stabilizing thiohemiacetal formation/accumulation and, thereby, affect K_i^* .

At this juncture time-dependent, slow binding inhibitors, phosphinates la-c and aldehyde 2 have been developed that selectively block either the Gsp synthetase $(1a-c)$ domain or the Gsp amidase (2) domain of the bifunctional E. coli Gsp synthetase/amidase enzyme. Each takes advantage of mechanistic pathway analysis to have each active site catalyze isomerization to the tightened complex.

Significance

The *Escherichia coli* glutathionylspermidine (Gsp) synthetase/amidase is an intriguing catalyst for at least two reasons. First, the physiological function of Gsp formation in E. coli from the antioxidant thiol glutathione and the polyamine spermidine is not yet understood. Although Gsp is a precursor to bis(glutathionyl)-spermidine (trypanothione) in trypanosomatid parasites, trypanothione has not been detected in E. coli. Second, the juxtaposition of an ATP-dependent amide synthetase and an opposing amidase in a bifunctional protein is unusual and requires regulation at the enzyme level to and future and supermitted and specification and supermitted and specific arba ramo grammono and sportmanic-acport ATPase action. The demonstration that γ -Glu-Ala-Gly aldehyde is a potent time-dependent specific inhibitor of the amidase activity supports an S-acyl-enzyme mechanis animalise activity supports and staty emby me meeting. f_{non} and f_{non} a fold modulation of amidase K_i^* down to a 80-250 nM range either by occupancy of the synthetase domain, by
its own specific slow-binding inhibitor, or by removal of

the synthetase domain entirely (amidase fragment), corroborate interdomain communication and negative regulation of the amidase by the synthetase domain at rest. With the y-Glu-Ala-Gly aldehyde for the amidase and the three Gsp phosphinate analogs (Ala, Ser, Cys) for the synthetase, potent mechanism-based slow-binding inhibitors specific for each active site of the bifunctional E. coli Gsp synthetaselamidase are available for dissection of mechanism and physiological role.

Materials and methods

Synthesis of phosphinate and aldehyde inhibitors and synthetase substrates

Experimental procedures for the synthesis of 1b, 1c and 2 may be found in the Supplementary material. Synthesis of the tripeptides y-Glu-Ala-Gly [6] and y-Glu-Ser-Gly for use as alternative synthetase substrates was carried out using conventional solution-phase techniques Experimental details for synthesis of the Ser-containing peptide are available from the authors (S.C. & J.K.C.).

Preparation of Gsp amidase fragment and full-length Gsp synthetaselamidase

Proteins were overproduced in E . coli and purified according to [3,6].

Gsp amidase inhibition assay by γ Glu-Ala-Gly CHO

The y-Glu-Ala-Gly-p-nitroanilide was used as substrate for inhibition of the Gsp amidase activity [8]. Assay mixtures of 390 ul contained 100 mM NaPlPES (pH 6.8) and 11.4 mM y-Glu-Ala-Gly-p-nitroanilide (24 mM used for the full-length Gsp synthetase/amidase). Gsp amidase fragment (10 μ l of 256 nM) (1.2 μ M of the full-length protein) were added to initiate inhibition assays. The absorption at 405 nm $(\epsilon = 9940 \text{ cm}^{-1} \cdot \text{M}^{-1})$ and 37°C was monitored to measure the release of p-nitroaniline.

The resulting data were analyzed and determined to follow a slowbinding inhibition pattern with an isomerization from the initial enzyme-inhibitor complex E.1 to the final tight-binding complex E.l* (equation 1). The kinetics can be described on the basis of an integrated equation (equation 4), in which V_0 and V_s are the initial and the steady-state rates, k is the apparent rate constant for establishing the steady-state equilibrium, and P is amount of product accumulated during a period of time t.

$$
p = V_s \cdot t + (V_o - V_s) \frac{(1 - e^{-kt})}{k}
$$
 (4)

 V_{o} and V_{s} are defined as in equations 5 and 6: where K_{i} and K_{i}^{*} are the initial and final inhibition constants, respectively.

$$
V_o = \frac{V_{max} \cdot [S]}{K_m (1 + \frac{[I]}{K_i}) + [S]}
$$
(5)

$$
V_s = \frac{V_{max} \cdot [S]}{K_m(1 + \frac{[I]}{K_s}) + [S]}
$$
(6)

Gsp synthetase inhibition assay by phosphinates la-c U sp synthetase inhibition assay by phosphinates I a-c

 r_{in} and r_{in} by r_{in} and r_{out} and r_{out} are r_{out} of r_{out} of r_{out} v_{c} includes the dependix of σ as in the assay was interested by v_{c} was interested by v_{c} vate kinase/lactate dehydrogenase reactions. The assay was initiated by adding purified Gsp synthetase/amidase (12.8 nM) to an assay mixture which contained the following components (final concentration):

1.56 mM glutathione, 10 mM spermidine, 2 mM ATP, 2.7 mM MgCl₂, 1 mM phospho(enol)pyruvate, 0.2 mM NADH, 50 ug ml⁻¹ lactate dehydrogenase, 100 μ g m $^{-1}$ pyruvate kinase, and various concentrations of inhibitor all in 50 mM NaPlPES (pH 6.8) at 37°C. Slow binding inhibition with $E_1 \rightleftharpoons E_1$ ^{*} was also observed for the synthetase inhibition, and equations 3-5 were also used for analysis to calculate K, and K * .

Dissociation of E/t^* complex

The E.l* complex of Gsp amidase fragment and aldehyde 2 was prepared by incubation of 120 nM Gsp amidase fragment with 200 uM γ -Glu-Ala-Gly·CHO at room temperature for 20 min. A sample (3 µI) of the incubation mixture was withdrawn and diluted 1 OO-fold in 300 ul of 100 mM Na-PIPES (pH 6.8) containing 36 mM y-Glu-Ala-Gly-pNA as the assay substrate. Recovery of Gsp amidase activity was assayed at 37°C by monitoring the release of p-nitroanilide at 405 nm (light path = 2 mm). The resulting data were transferred into KaleidaGraph and fitted by equation 7 (p: product formation; V_s : steady state rate) to determine the dissociation constant $k₄$ (see equation 1 for definition).

$$
p = V_s \left(t - \frac{1}{k4} \left(1 - e^{-k4t} \right) \right) \tag{7}
$$

The E-I* complex of Gsp synthetase and the phosphinate (1a, 1b or 1c) was prepared by incubation of the full-length Gsp synthetase/amidase enzyme (100 nM) with 2 uM phosphinate inhibitor, 2 mM ATP and 2 mM MgCl₂ at room temperature. After 30 min, 5 ul of the incubation mixture was withdrawn and diluted 1 OO-fold in 500 ul of 100 mM Na-PIPES (pH 6.8) containing 8 mM glutathione, 10 mM spermidine, 2 mM ATP, 2.7 mM MgCl₂, 1 mM phospho(enol)pyruvate, 0.2 mM NADH, $50 \,\mu g$ ml⁻¹ lactate dehydrogenase and $100 \,\mu g$ ml⁻¹ pyruvate kinase. Recovery of Gsp synthetase activity was assayed at 37°C. Equation 7 was used to determine the dissociation constant and the half-life of E-l* complex.

Supplementary material

Experimental details for the synthesis of compounds 1b, 1c and 2 are published with this paper on the internet.

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